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a) It has only sigma bonds. b) It has only pi bonds. c) It has a sigma and two pi bonds. d) It has a sigma and delocalized pi bond. [View Answer.](#)

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Answer: d. **Explanation:** An aromatic hydrocarbon always has a sigma as well as a delocalized pi bond found between the carbon atoms.
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Choose the answer that best fills in
the blanks of the following

statement: An aromatic must have

_____ of pi electrons, above and below

the _____ of the molecule. Tutor Me

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question complexity. Median response time is 34 minutes and may be longer for new subjects. Q: Determine the standard potential, E, of a cell that employs the reaction: $\text{Fe} + \text{Cu}^{2+} \rightarrow \text{Cu} + \text{Fe}^{2+}$. - Rea... A: The difference of potential values of the half cells (E ...

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Answered: Which of these is an aromatic compound?... | bartleby
Yes, it is aromatic. $4n+2$ pi-electrons.
S15.6.2 There is only one nitrogen of the core that contributes to the pi-system (in red). With this one lone pair the core is aromatic with 10 electrons in the pi-system.

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Question 3. Which of the following statements is incorrect: aromatic compounds... a) Are planar. b) Have $4n + 2$ -electrons. c) Are cyclic. d) Are generally less reactive than similarly substituted alkenes.

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Q1: Histidine pictured below is
considered a basic amino acid due to
WHICH nitrogen (blue or green)
atom? Explain your answer. Bonus Q2:
Explain why [16]Annulene pictured

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1. Huckel rule 's: For a compound to be aromatic, following conditions must be followed. (i) Compound must be cyclic & planar or nearly planar. (ii) It must show resonance &

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delocalization. (iii) It obeys $[4n + 2]$ electron rule: n is called Huckel ' s number or aromaticity index.

Aromaticity index $[n] = 0, 1, 2, 3, \dots$

Aromaticity - General Organic
Chemistry Chemistry Notes ...

Aromatic hydrocarbons are

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Aromatic monocyclic or polycyclic. Most of the aromatic compounds contain a benzene ring. They are also known as arenes. In aromatic hydrocarbons, pi electrons are delocalised between carbon atoms in the cycle.

Heterocyclic compounds following Huckel ' s rule are also aromatic

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Answers. Benzene is the most popular aromatic hydrocarbon.

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This aromatic hydrocarbons (benzene

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