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1. 2. -2- Which of the following pairs of formulae is true for butanoic acid?

Molecular Empirical Formula

(A) $C_4H_8O_2$ CH_2O (B) $C_4H_6O_2$ CH_2O (C) $C_4H_8O_2$ C_2H_4O (D) $C_4H_8O_2$ C_2H_4O

Combustion analysis of 0.18 g of an organic compound produces 0.396 g of carbon dioxide and 0.216 g of water.

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SOLUTION AND EXPLANATIONS (a)

A dative covalent or coordinate covalent bond is one in which one of the atoms supplies both the shared electrons to the covalent bond. A dative covalent bond is written $A \rightarrow B$, where the direction of the arrow shows the direction in which the electron pair is donated.

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