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the Periodic Of Table

Gram Positive vs. Gram

Negative
Bacteria

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ks/Rules/Concept

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Arrangement Of

Modern Chemistry

1 Arrangement of

Electrons in

Atoms CHAPTER 4

REVIEW

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Arrangement of Electrons in Atoms Teacher Notes and

Answers Chapter

4 SECTION 1

SHORT ANSWER 1.

In order for an electron to be ejected from a metal surface, the electron must be struck by a single

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Arrangement Of
photons with at
least the
minimum energy
needed to knock
the electron
loose.

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Electrons in
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Atoms . . . Holt

Modern Chemistry

Review CHAPTER

4: ARRANGEMENT

OF ELECTRONS IN

ATOMS. The

following pages

contain the bulk

(but not all) of

the information

for the chapter

4 test. Focus on

this content,

but make sure to

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review class
notes,
activities,
handouts,
questions, etc.

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Transcription. 1
CHAPTER 4 PRE-
TEST Arrangement
of Electrons in

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Atoms In the

space provided,
write the letter
of the term that

best completes
each 3 Chapter 4
PRE TEST,

continued 8.

What is the
frequency of a
photon whose
energy is J ? (h
 $= J \text{ s}$) a Hz b Hz
c Hz d Hz 9.

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When electromagn etic...

Electrons In

Atoms

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In the space provided, write the letter of the term that best completes each sentence or best answers each question.

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Modern Chemistry
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electrons
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Electrons In *Atoms Chapter 4* *Test Answers*

CHAPTER 4 REVIEW

Arrangement of
Electrons in
Atoms SECTION 2
SHORT ANSWER

Answer the
following
questions in the
space provided.

1. d How many
quantum numbers

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are used to

describe the
properties of
electrons in

atomic orbitals?

(a) 1 (c) 3 (b)
2 (d) 4 2. a A

spherical

electron cloud

surrounding an

atomic nucleus

would best

represent (a) an

s orbital.

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*4 Arrangement of
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28 Chapter Test
Name Class Date
Chapter Test A,
continued _____

8. What is the frequency of a photon whose energy is 3.4×10^{-19} J? ($h = 6.626$

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10³⁴ J·s) a.
8.8 × 10²⁶ Hz b.
5.1 × 10¹⁴ Hz c.
1.9 × 10⁻¹⁵ Hz d.
2.3 × 10⁻⁵² Hz

_____ 9. When electromagnetic radiation strikes the surface of a metal, electrons are ejected from the metal's ...

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Assessment Of

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Atoms . SHORT
ANSWER Answer
the following
questions in the
space provided.

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1. How many quantum numbers are used to describe the properties of electrons in atomic orbitals?

- (a) 1 (b) 2 (c) 3 (d) 4

 A spherical electron cloud surrounding an atomic nucleus

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electrons in
atoms. The
attractions
between the

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protons and
electrons of
atoms can cause
an electron to

move completely
from one atom to
the other. When
an atom loses or
gains an

electron, it is
called an ion

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4-3 The Arrangement Of
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effect:

electromagnetic
radiation
strikes the
surface of the
metal, ejecting
electrons from
the metal and
creating an
electric
current.

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wavelength are
mathematically
related to each
other.

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