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the molar mass of C3H4? c. How many moles are in an 8.0 g sample of C3H4? 2. a. What ...

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actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N 2 are mixed with 12.0 mol of H 2 according to the following equation: N 2(g) 3H 2(g ...

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the coefficient x in this equation? 40.07 g/mol h. What is the molar mass of C3H4? 2 mol 02:1 mol H20 c. What is the mole ratio of 02 to H20in the above equation? 0.20 mol d. How many moles ...

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