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General Chemistry 1 Review

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KEY 1 REVIEW QUESTIONS

Chapter 9 1. Draw Lewis structures for each of the following structures and assign formal charges to each atom: a) SF₂ 2 20 electrons 0 0 0 b) NH₂ OH (N and O are bonded to one

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another) 14 electrons 0 0 0
0 0 c) PO₄³⁻

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Chemistry 101 ANSWER KEY 1

REVIEW QUESTIONS Chapter 10

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1. Draw Lewis structures and determine the molecular geometry of each molecule or ion shown below: A) ClO_2^- 20 electrons Bent (from tetrahedral) B) ICl_3 28 electrons T-shaped (from trigonal bipyramidal) C) TeF_4

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4 34 electrons See-saw (from
trigonal bipyramidal) 2.

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Chemistry 101 ANSWER KEY 1

REVIEW QUESTIONS Chapter 8

Use only a periodic table to
answer the following

questions. 1. Write complete
electron configuration for

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Review of the following

elements: a) Aluminum (Al)

$1s^2 2s^2 2p^6 3s^2 3p^1$ b)

Sulfur (S) $1s^2 2s^2 2p^6 3s^2$

$3p^4$ c) Manganese (Mn) $1s^2$

$2s^2 2p^6 3s^2 3p^6 4s^2 3d^5$ 2.

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REVIEW OF BASIC CHEMISTRY

ANSWER KEY 1. Name the following elements. Spelling counts: H hydrogen S sulphur Cl chlorine C carbon N nitrogen Na sodium Pb lead Ag silver 3. Using a

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periodic table, record the
atomic number for the
following elements:

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ANSWER KEY 1 REVIEW

QUESTIONS Chapter 8 Use only

a periodic table to answer

the following questions. 1.

Write complete electron

configuration for each of

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the following elements: a)
Aluminum (Al) $1s^2 2s^2 2p^6$
 $3s^2 3p^1$ b) Sulfur (S) $1s^2$
 $2s^2 2p^6 3s^2 3p^4$ c) Manganese
(Mn) $1s^2 2s^2 2p^6 3s^2$

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Use only a periodic table to
answer the following
questions. 1. Write complete

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electron configuration for
each of the following
elements: a) Aluminum (Al)
1s² 2s² 2p⁶ 3s² 3p¹ b)
Sulfur (S) 1s² 2s² 2p⁶ 3s²
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1. Draw Lewis structures and

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determine the molecular geometry of each molecule or ion shown below: A) ClO_2^- 20 electrons Bent (from tetrahedral) B) ICl_3 28 electrons T-shaped (from trigonal bipyramidal) C) TeF_4 34 electrons See-saw (from

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Chapter 1 Review Questions

1) Molecules can be described as A) mixtures of two or more pure substances. B) mixtures of two or more elements that has a specific ratio between components. C)

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two or more atoms chemically joined together. D) heterogeneous mixtures. E) homogeneous mixtures.
Answer: C 2) Identify the TRUE statement.

General Chemistry 101

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chemistry. study of the
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the changes it undergoes.
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that contain carbon.
inorganic chemistry. study
of essentially all compounds
that do not contain carbon.
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