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Principle states that

when a system at

equilibrium is

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subjected to a stress,
the system will shift its
equilibrium point in
order to relieve the
stress.

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Name Classify the

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as ionic (metal +

nonmetal), covalent

(nonmetal + ...

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5. 35 g of

$\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

Determine the
number of grams in
each of the quantities

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NaCl 2. 0.50 moles of
3, 1.70 moles of
KMnO₄ 4, 0.25 moles
of KCl e, ST 5. 3.2
moles of
CuSO₄·5H₂O
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relation concerning
the compression and
expansion of a gas at
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