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Consider your answers to questions 5 and 6. Were there any that gave the same answer (or just about the same answer) If so, which ones 5c and 6c: $\text{pH} + \text{pOH} = 14$ 5d and 6d: $[\text{H}^+] \times [\text{OH}^-] = 1.0 \times 10^{-14}$ 8. Given your answer to number 7, you should be able to answer the following questions.

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PLEASE NOTE: If you have a question about these answers, it is your responsibility to come to office hours or ask during class work time. UNIT 12 - HW Practice Keys - ChemActivity 51: Cell Voltage - ChemActivity 50: Electrochemical Cell UNIT 11 - HW Practice Keys - ChemQuest 55: Free Energy - ChemQuest 54: 2nd Law of Thermodynamics

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Information: Relating Electron Configurations to the Periodic Table
In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready. Remember that a row on the periodic table

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