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 Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q1 and Q2 separated by a distance r is given by QQ PE = 4?? o r First we consider the interaction between Na+ ion and 6Cl? ions at distance r Applying Coulomb's law we have Q1Q2 (?6e)(+e) ? 6e PE1 = = = 4?? o r 4?? o r 4?? o r Similarly, we now consider 12 Na+ ions as second nearest neighbors at a distance r QQ (+12e)(+ e) 12e = PE2 = = 4?? o r 4?? o r 4?? o r and Cl ...

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 Solutions to Principles of Electronic Materials and Devices: 4th Edition (25 April 2017)Chapter 11.2 Virial theorem and the He atom In Example 1.1 we calculated the radius of the H-atom using theVirial theorem. First consider the He+ atom, which as shown in Figure 1.75a, has one electron in the Kshell orbiting the nucleus.

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