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## ~~Flinn Chemtopic Labs Answers~~

$\text{NaCl(aq)} + \text{AgNO}_3\text{(aq)} \rightarrow \text{AgCl(s)} + \text{NaNO}_3\text{(aq)}$   
 $\text{H}_2\text{SO}_4\text{(aq)} + 2\text{NaOH(aq)} \rightarrow \text{Na}_2\text{SO}_4\text{(aq)} + 2\text{H}_2\text{O(l)}$   
 $\text{N}_2\text{S}_3\text{(aq)} + 2\text{HCl(aq)} \rightarrow 2\text{NaCl(aq)} + \text{H}_2\text{(l)} + \text{SO}_2\text{(g)}$   
A compound burns in the presence of oxygen, producing energy in the form of heat and light. The combustion of organic compounds produces carbon dioxide and water.

## ~~Classifying Chemical Reactions – 50 Webs~~

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Summary. In this inquiry-based experiment, the students design and carry out a sequence of chemical reactions for the separation and identification of iron, silver, and zinc ions in water. Two parallel series of tests will be carried out, one using a known sample containing all three metal cations and the other using unknown samples containing one, two, or all three cations.

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