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Mass = 0.200 L x 1.14 mol C₇H₅O₂ / 1L x 1mol NaC₇H₅O₂ / 1 mol C₇H₅O₂ x 144 g NaC₇H₅O₂ / 1 mol NaC₇H₅O₂ = 32.832 g NaC₇H₅O₂ References Petrucci, et al. General Chemistry: Principles & Modern Applications. 9th ed. Upper Saddle River, New Jersey 2007.

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It is equivalent to General Chemistry I/II (1403/1404) taught during the regular academic year. The first term of the course is comprised largely of (1) a brief review of introductory chemistry concepts, followed by (2) atomic and molecular structure and bonding in depth and (3) a comprehensive survey of organic and inorganic chemistry.

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