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The name of a base is the name of the cation followed by the name of the anion (hydroxide). The formula of a base is written by showing the number of hydroxide ions needed to balance the positive charge on the cation. After reading Lesson 9.4, answer the following questions.

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Names and Formulas of Acids hydrogen1.

## **Chemical Names and Formulas - Mr. Mutic's Chemistry ...**

a.potassium sulfidee.hydrobromic acidf.sulfur hexafluoride. b.tin(IV) chloridef.aluminum fluoridej.magnesium chloride. c.hydrosulfuric acidg.dinitrogen pentoxidek.phosphoric acid. d.calcium oxideh.iron(III) carbonatei.nitric acid. Name

\_\_\_\_\_ Date \_\_\_\_\_ Class

\_\_\_\_\_.

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charges are anions. The names of nonmetallic anions end in -ide.

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Metallic cations take the name of the metal. Some metals, including transition metals, can form more than one cation. Use a Roman number in the Stock name or use the classical name with a suffix to name these metals. 1

## **Chemical Names and Formulas - Pittsfield**

This is a list of the 118 chemical elements which have been identified as of 2020. A chemical element, often simply called an element, is a species of atoms which all have the same number of protons in their atomic nuclei (i.e., the same atomic number, or Z).. A popular visualization of all 118 elements is the periodic table of the elements, a convenient tabular arrangement of the elements by ...

## **List of chemical elements - Wikipedia**

PEARSON CHEMISTRY: The law of conservation of mass is explained in Lesson 2.4 (p. 50). Students are introduced to balancing chemical equations in Lesson 11.1 (pp. 346- 354). Students learn to write and balance the five different types of chemical reactions in Lesson 11.2 (pp. 356-367). Students learn

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