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~~Calculate The  
Number of Protons,  
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Chemistry - Atomic  
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Protons Neutrons

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Structure of Atom

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Atomic Number,  
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Kids will learn about some important subatomic particles, and then they'll get to try a simple experiment to see atoms in action.

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Atomic Structure  
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3 subatomic  
particles, their  
charges, masses  
and where in the  
atom they are  
located in the chart  
below: Describe  
how to determine

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the number of each subatomic particle in an atom below  
Fill in the chart for the elements below. Round all atomic masses to the nearest whole number: Draw an atom below and label

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The protons and

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neutrons are at the center of the atom, the nucleus, with empty space around it.

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\*Here's how to draw an atom of any element using the Bohr model of the atom: Protons

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and neutrons go in  
the nucleus  
(center). Electrons  
go in energy levels  
of the electron  
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Marks : \_\_\_\_\_ A.

Structure of an

atom 1. Fill in the

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blanks concerning  
sub-atomic  
particles. (a) There  
are three sub-  
atomic particles in  
atoms. They are \_  
protons \_, \_  
neutrons \_ and \_  
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## Answer Key

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tells you the  
number of \_\_\_\_\_ in  
a neutral atom of  
that element. The  
atomic number

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gives the “identity” of an element as well as its location on the periodic table. No two different elements will have the \_\_\_\_\_ atomic number. 4. The \_\_\_\_\_ of an element is the average mass of an element’s naturally occurring atom, or

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worksheet with all

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E.g. details of

subatomic

particles, mass &

atomic numbers.

Great activity for

revision.

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Name three sub atomic particles of an atom. 2. Define atomic number and mass number. 3. Describe the particle scattering experiment conducted by Rutherford with a

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Atom

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of the Atom is  
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provided for

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practice and self  
evaluation of  
students of class 9.  
Solutions are  
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Write electronic  
configuration of  
element X which  
has atomic number  
= 15. Write its  
valency also.

~~Structure of the~~

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General Chemistry

Mr. MacGillivray

Atomic Structure

Worksheet 1. The  
number of protons  
in the nucleus of an  
atom of a

particular element

is called the \_\_\_\_\_

\_\_\_\_\_ of that

element. 2. In a

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neutral element,  
the number of  
protons must equal  
the number of  
3.

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