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TEACHING TRANSPARENCY WORKSHEET Ionic Bonds Use with Chapter 7, Section 7.2

1. How many valence electrons does a neutral magnesium (Mg) atom have?
2. What is the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so?
3. How many valence electrons does a single neutral chlorine (Cl) atom have?
- 4.

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uest 11 Name: Date: Hour: Information: Energy Levels and Sublevels ... Considering your answer to question 5, how many electrons can fit in the entire 5th energy level? can (A₅ is a ha AC5a) 50đ 7. Write the notation for an electron spinning clockwise in a p sublevel ...

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1. For each of the following pairs of elements, underline the one that would replace the other element in a compound.
a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium
- 2.

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TEACHING TRANSPARENCY WORKSHEET The Activity Series

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transparency, worksheets, chemistry, answers, chapter, 19 Created
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Chapter 8 . 1.7, the CF bond is polar covalent. (b) $\text{EN}_{\text{Cl}} - \text{EN}_{\text{P}} = 3.0 - 2.1 = 0.9$. Since $0.5 < 0.9 < 1.7$, the PCl bond
is polar covalent. (c) $\text{EN}_{\text{Cl}} - \text{EN}_{\text{Li}} = 3.0 - 0.9 = 2.1$. Since $2.1 > 1.7$, the PCl bond
is ionic. Section 8.5 Electronegativity and
Polarity 265 Compare and contrast polar . that is in excess.

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Answer Key Teaching Transparency 11 - Teacher Worksheets

TEACHING TRANSPARENCY WORKSHEET Atomic Orbitals 1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2 2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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